INTERNAL ASSESSMENT - II CHEMISTRY

Full Marks: 20 Class-VIII Time: 45 minutes

Attempt all questions

1. [1x3=3]

i. What was the thickness of gold foil used in Rutherford's experiment?

- ii. Write the electronic configuration of Calcium(Z=20).
- iii. How many electrons can be accommodated in the fourth (or N) shell of an atom?

2. [2x4=8]

- i. Mention two main differences between a mixture and a compound.
- ii. The atomic numbers of Al and Cl are 13 and 17, respectively. What will be the numbers of electrons in Al³⁺ and Cl⁻?
- iii. What are the postulates of Thomson's Atomic Model?
- iv. Which separation technique will you apply for the separation of the following?
 - (a) Wheat grains from husk
 - (b) Butter from curd

3. [3x3=9]

- i. Draw a flow-sheet describing separation of the constituents of a mixture containing aluminium powder, iron filings and sugar.
- ii. Describe sedimentation and decantation with the help of an activity. What kind of mixtures can be separated by decantation?
- iii. Electronic configuration of an atom X is (2, 8, 7). It combines with another atom Y having electronic configuration (1):
 - (a) What type of bond will be formed between X and Y
 - (b) Write the formula of the compound
 - (c) Is it soluble in water?